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## **Acids Bases And Salts Answers**

A reaction in which an acid and base react with each other to give a salt and water is termed as neutralization reaction. For Example: (i)  $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$  (ii)  $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$

## **NCERT Solutions for Class 10th Science: Ch 2 Acids, Bases ...**

Answer-1. Post-Your-Explanation-1. 2. Bases react with acids to produce salts and water. water to produce acids and salts. salts to produce acids and water. neither acids, salts, nor water. Answer-2.

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## **Acids Bases and Salts Multiple Choice Questions and Answers**

Access Answers of Science NCERT class 10 Chapter 2 - Acids Bases and Salts. In-text questions set 1 —> Page number 18. 1. You are given three test tubes.

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## **NCERT Exemplar Class 10 Science Solutions Chapter 2 ...**

Class 10 Science Chapter 2 MCQ (Multiple Choice Questions) of Acids, Bases and Salts. All the MCQs are taken from NCERT Textbooks issued for

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session 2020-2021. Answers and explanation of MCQ tests are given just below each questions. MCQs are important for the CBSE and other boards final exams. It also clears the concepts about Acids, Bases and ...

## **Class 10 Science Chapter 2 MCQ of Acids, Bases and Salts ...**

Salt is formed by combination of acid and base through neutralization reaction. water of Crystallization : It is the fixed number of water molecules present in one formula unit of a salt e.g.

## **NCERT Solutions for Class 10 Science Chapter 2 Acids ...**

Topic 8 - Acids, Bases and Salts Notes || Questions by Topic This topic is included in Paper 1 , Paper 2 , Paper 3 , Paper 4 , Paper 5 and Paper 6 for IGCSE CIE Chemistry.

## **CIE IGCSE Chemistry Topic 8: Acids, Bases and Salts ...**

Answer. (a) HCl is acid and NaOH is base

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whose combination forms the common salt. Its formula is  $\text{NaCl}$  (Sodium chloride). It is obtained from sea water. (b) Rock salt is the common name for the mineral "halite". Its chemical formula is  $\text{NaCl}$ . It may be white or light blue or yellow depending upon impurities present in it.

## **Acids Bases and Salts Chapter Wise Important Questions ...**

Answer: Earlier, acids, bases, and salts were characterized by the experimental testing of their aqueous solutions. An acid is defined as a substance whose water...

## **What is Acids , Salts and Bases? - Brainly.in**

Play this game to review Chemistry. Which of the following values would represent the pH of a strong base?

## **Acids/Bases and Salts | Chemistry Quiz - Quizizz**

Short Answer type Question 10 Give a

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reason why – (a) Great care should be taken while handling laboratory acids and bases. (b) Solid powder of baking soda does not turn dry litmus paper to blue in color. Question 11 Define acid rain and how it is produced? What are its harmful effect? Question 12 Define neutralization reaction with an example.

## **Worksheet for Class 7 Science Chapter 5: Acid Base and Salt**

(b) Acids (c) Bases (d) Salt Q6: Which of the following is a strong acid? (a) acetic acid (b) nitric acid (c) lactic acid (d) citric acid Q7: The most commonly used natural indicator 'Litmus' is extracted from (a) Lichens (b) Turmeric (c) Beet Root (d) China rose Q8: Which of the following colours is given by phenolphthalein in a basic salt ...

## **CBSE Class 7 - Science - CH5 Acids, Bases and Salts (MCQs)**

class 7, chapter 5, science question answer and problem solving, can ask question ans in comment section, will

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revert as soon as possible. ... science, chapter 2, acid , base & salt. - duration ...

## **CLASS 7, CHAPTER 5, SCIENCE , ACIDS BASES AND SALTS**

i.e. acid + base → salt + water

$\text{HCL} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$  The salts are formed as a result of neutralization reaction as both acid and base neutralize each others effect.

## **Acids, Bases and Salts Class 10 Science, Explanation ...**

Name the salt formed when hydrochloric acid reacts with sodium hydroxide solution (a base). Answer: When hydrochloric acid reacts with sodium hydroxide solution, then sodium chloride salt is formed.

## **Acids, Bases and Salts Class 7 Extra Questions Science ...**

In this article, we are providing you the MCQs from CBSE Class 10 Science Chapter 2: Acids, Bases and Salts. All the



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