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waves. $c = 3 \times 10^8$ m/s for all electromagnetic waves. amplitude amplitude
A B Interpreting Waves MATH SKILLS TRANSPARENCY MASTER
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Transparency Masters Chemistry: Matter and Change • Chapter 11 31 number of moles of known substance C H C H C H O O O CO CO CO H O H O H O number of moles of unknown substance O CO H O C H CO H O C H O H O C H O CO use mole ratio 5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...

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MATH SKILLS TRANSPARENCY MASTER. Use with Chapter 4, Section 4.3. 4. Given the data in the table, calculate the atomic mass of unknown element X. Then, identify the unknown element, which is used medically to treat some mental disorders. Analyze the Problem. You are given the data in the table.

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Math Skills Transparency Worksheets Chemistry: Matter and Change • Chapter 6 41 1. Identify the number of valence electrons in each of the following elements. a. Ne e. O b. K f. Cl c. B g. P d. Mg h. Si 2. Identify the energy level of the valence electrons in each of the following elements. a. Br b. N c. Ra d. H e. Ar f. I 3.

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Use with Chapter 10, Section 10.3 MATH SKILLS TRANSPARENCY WORKSHEET Calculating the Molar Mass of a Compound

Determine the molar mass of each of the following compounds.

Use the periodic table. 1. carbon dioxide (CO_2) \ 2-0 002) 2.

mercury(I) fluoride (Hg_2F_2) 3. magnesium thiotellurite

(Mg_3TeS_5) 32.1 4. copper(II) cyanide o

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MATH SKILLS TRANSPARENCY WORKSHEET Determining Numbers of Ions Use with Chapter 7, Section 7.3 1. Write a simple word equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound. 2. How many potassium ions (group 1) would be related to balance the charge of each of

Math Skills Transparency Worksheet Answers Chapter 9

MATH SKILLS TRANSPARENCY WORKSHEET Use with Chapter 25, Section 25.3 42 82 1. In equation 1, what do the numbers 238 and 92, written to the left of the symbol U, represent? 238 is the mass number of the isotope, and 92 is the atomic number of the isotope. 2. Explain how both mass number and atomic number have been conserved in equation 1.

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