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If the ratio  $h_{fg}$  to  $TV$  is assumed approximately constant, then

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$P_{\text{sat}} = A + B/T$  If the ratio  $H_{\text{sl}}$  to  $V_{\text{sl}}$  is assumed approximately constant, then  $P_{\text{sat}} = A + B \ln T$ .  
6.99 By Eq. (6.73) and its analog for sv equilibrium:  $dP_{\text{sat}} = P_{\text{t}} \frac{H_{\text{tsv}}}{R T^2} dT$   
 $Z_{\text{tsv}} \frac{R T^2}{P_{\text{t}}} \frac{dP_{\text{sat}}}{dT} = P_{\text{t}} \frac{H_{\text{tsv}}}{R T^2} dT$  ...

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Manual Solution Manual Thermodynamics Moran Shapiro 5th. 00 mol CH 4 16. Chapter 1 Getting Started: Introductory Concepts and Definitions. \*\*\* Note that  $h_N(t) = X_{N1} N_{1P}(N_1; t)$ : Using the stochastic equation, it is easy to see that  $h_N(t + \tau) = 1 - 2 N h_N(t + 1)$ ; which leads to.

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